

Evaluation of 2,3-Disubstituted Quinazolones for their Antimicrobial Study by the Standardized Procedures as Recommended by the National Committee for Clinical Laboratory Standards (NCCLS)

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Abstract

Heating under reflux a mixture of anthranilic acid and benzoyl chloride in dry pyridine afforded 2-phenyl-4H-benzo [d] [1,3] oxazine-4-one(1). Reaction of (1) with an excess equivalent of hydrazine hydrate in dry pyridine solvent furnished 3-amino-2-phenylquinazolin-4(3H)-one(2) which on reaction with carbamide in ethanol containing catalytic quantity of HCl yielded 1-(4-oxo-2-phenylquinazolin-3(4H)-one)(3). Interaction of (3) with arylamido/ imidoalcanolsin conc. H₂SO₄ generated targets compounds(4) designated as N- {[3-(4-oxo-2-phenylquinazolin-3(4H)-yl)ureido] alkyl}arylamides/imides and were evaluated for their antimicrobial activity against four bacterial and four fungal strains.

Introduction

Literature survey reveals that the quinazole compounds have been extensively investigated in several health areas [1–8]. Interest in quinazoles has increased manifolds because of their association with anticancer activity. The antitumour properties of some such compounds have been reported in culture of L1210 cells [4]. Recently, several research papers describing the antiviral and anti-microbial activities of quinazoles have appeared. Thus, the quinazolyl benzimidazoles were found antivirally active against Japanese Encephalitis virus (JEV), a highly pathogenic virus afflicting mostly children with high mortality rate and Herpes simplex virus (HSV-I). Low to moderate order of antiviral activity was observed against both the animal viruses. Quinazole derivatives containing isoquinoline and thiadiazole nuclei were found to exhibit antiviral and antifungal properties.

These compounds were synthesized for their antifungal activity against *Fusarium solanis* a casual organism of Guava Wilt Disease (GWD). One quinazole compound of this category was found to show percent inhibition of fungal zone to the extent of 70% while others showed low order of antifungal activity [9]. In addition, thiadiaquinazolones were found active against JEV and HSV-I *in vitro* [10]. Thus, it is evident that the quinazole compounds find greater applications in several health areas. The significant diverse pharmological properties exhibited by quinazole compounds led the authors to undertake the synthesis of N-{[3-(4-oxo-2-phenylquinazolin-3(4H)-yl] ureido]alkyl} arylamides/imides for their antimicrobial activity (4) involving four bacterial and four fungal strains by the standardized methods as recommended by the National Committee for Clinical Laboratory Standards (NCCLS) (Figure).

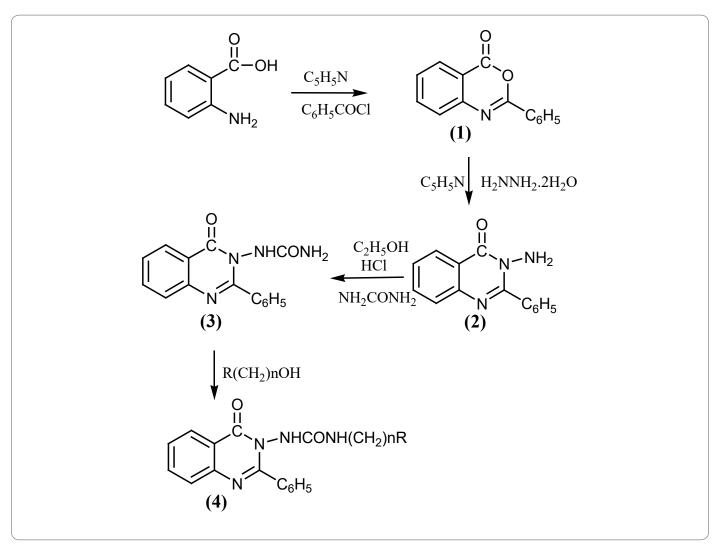


Figure: Scheme.

Experimental Section

1-(4-oxo-2-phenylquinazolin-3(4H)-urea) (3)

3-Amino-2phenylquinazoline-4-(3H)-one (2) (0.05 mole) and urea (carbamide) (0.05 mole) in absolute ethanol containing catalytic amount of concentrated hydrochloric acid (2 ml) were heated under reflux for two hours in such a manner that the reaction mixture was free from atmospheric moisture. Subsequently, solvent was removed by distillation and after washing with cold water, was dried at 100°C. It was recrystallized from acetone as light brown crystalline mass. The analytically pure sample melted at 180°C to 181°C, yield 81%.

N- {[3-(4-oxo-2-phenylquinazolin-3(4H)-yl) ureido] alkyl} arylamides/imides (4)

1-(4-Oxo-2-phenylquinazolin-3(4H)-urea) (3) (0.02 mole) and an electrophilic reagent; N-(hydroxy alkyl compound) viz; N-(hydroxy methyl) phthalimide/ N-(hydroxy methyl)- benzamide/N-(hydroxy methyl)-nicotinamide/N-(hydroxy ethyl) phthalimide/ 2-phenylquinazolin-4-(3H)-one (0.02 mole)were finely powdered by grinding together and dissolved in concentrated sulphuric acid by stirring vigorously and carefully [11]. While dissolving, the contents were occasionally cooled so that the reaction temperature did not exceed 10°C. A dark solution resulted which was subjected for

further stirring mechanically. The acidic solution thus obtained was refrigerated overnight and subsequently poured in ice cold water (~250 ml) slowly with constant stirring. Precipitated solid mass was allowed to settle down and separated by filtration. The solid thus obtained, was washed successively with cold water in order to remove any sulphonated product and dried *in vacuo* overnight. It was recrystallized from diluted ethanol using animal charcoal as the decolorizing agent. The characterization data of the compound thus synthesized are recorded in (Table 1).

Biological Activity

Antimicrobial activity was performed as per the literature method [12–14].

The antimicrobial activity data of the investigational compounds are recorded in (Table 2).

Materials and Methods

For antibacterial activity, testing was done in peptone broth and for antifungal activity, the experimental compounds were tested employing the tube dilution technique. A 1.0 mg/ml solution of the tested compound was obtained by the dissolution in dimethyl sulphoxide solvent.

Com- pound No.	R	n	m.p.⁰C	Colour	Yield (%)	Molecular Formula	Molecular weight	I.R (KBr) in υ cm-1	1HNMR (CDCl ₃)	Mass (ES)
1	Phthalimido	1	165	Brown	65	C ₂ 4H ₁₇ N ₅ O ₄	439	1644 (C = N), 1665 (C = O) (sec. amide), 2962 (C-H str. in CH ₂)	7.14-7.69(m, 13H, ArH) 8.15 (brs, 2H, NHCO) 4.20(s, 2H, CH ₂)	M ⁺ 439, 396, 342, 314, 236, 133, 105 (Base peak)
2	Benzamido	1	170–171	Light brown	62	C ₂₃ H ₁₉ N ₅ O ₃	413	1709 (imide C = 0), 1646 (C = N),1670 (sec. amide C = 0), 1710 (imide C = 0), 2956 (C-H str. in CH ₂)	7.0-7.40 (m, 14H, ArH) 8.20 (brs, 2H, NHCO) 4.16 (s, 2H, CH ₂)	M ⁺ 413, 336, 325, 308, 297,177, 105, 77 (Base peak)
3	Phthalimido	2	170	Brown	59	C ₂₅ H ₁₉ N ₅ O ₄	453	1646 (C = N), 1685 (sec. amide C = 0), 2990 (C-H str. in CH_2), 1706 (imide C = 0)	7.16-7.70(m, 13H, ArH) 8.20 (brs, 2H, NHCO) 4.18 (s, 2H, CH ₂ CH ₂ symmetrical)	M+ 453, 376 332, 322, 297, 177, 105, (Base peak) 77, 42
4	Nicotinomido	1	160-161	Brown	60	C ₂₁ H ₁₈ N ₆ O ₃	402	1650 (C = N), 1690 (sec. amide C = 0), 2985 (C-H str. in CH_2), 1700 (imide C = 0)	7.06 -7.86 (m, 12H, ArH) 8.10 (brs, 2H, NHCO) 4.50 (s, 2H, CH ₂)	M ⁺ 402, 374, 248, 323, 295, 120, 77
5	3-[2- (phenyl)- quinazoline- 4-(3H)-one	2	159-160	Light yellow	55	C ₃₁ H ₂₄ N ₆ O ₃	528	1640 (C = N), 1670 (sec. amide C = 0), 2986 (C-H str. in CH ₂), 1698 (imide C=0)	7.06 -7.80 (m, 18H, ArH) 8.20 (brs, 2H, NHCO) 4.18 (s, 2H, CH ₂ CH ₂ symmetrical)	M ⁺ 528, 500, 451,307, 249, 221, 193, (Base peak), 114, 77, 43

	Table 1: Characterization data of	3-(4-oxo-2-phenylquinazolin-3(4H)-yl) ureido] alky	} arvlamides/imides.
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All the above compounds stowed satisfactory elemental (C, H and N) analyses.

Table 2: Antimicrobial activity of	lata of N- {[3-(4-oxo-2-phenylquinazo	blin-3(4H)-yl) ureido] alkyl} arylamides/imides.

Compound no.	Antibactirial ac	tivity data minim	um inhibitory co	ncentration	Antifungal activity data minimum inhibitory concentration				
	(µg/ml)				(µg/ml)				
	Ec	Pa	Sa	Кр	Са	Cn	Af	Ср	
1	50	> 50	> 50	> 50	> 50	> 50	25	50	
2	50	> 50	> 50	> 50	> 50	> 50	25	> 50	
3	50	> 50	> 50	> 50	> 50	> 50	> 50	> 50	
4	12.5	> 50	> 50	> 50	> 50	> 50	> 50	> 50	
5	25	> 50	> 50	> 50	> 50	> 50	> 50	> 50	

Ec: Escherichia Coli; Pa: Pseudomonas Aeruginosa; Sa: Staphylococcus Aureus; Kp: Klebsiella Pneumoniae; Ca: Candida Albicans; Cn: Cryptococcus Neoformans; Af: Aspergillus Fumigatus; Cp: Candida Parapsilosis.

The Minimum Inhibitory Concentration (MIC) of the investigational compound after incubating at 28°C was recorded after 72/96 hours post incubation.

Results and Discussions

The antimicrobial activity data reported in (Table 2) clearly indicate that all the five quinozolone compounds displayed some measurable degree of antibacterial activity against only one strain of bacteriai.e., against Escherichia coli (Ec). These quinozolone derivatives could not provoke any measurable level of antibacterial

activity against other bacterial strains. Quinozolone compounds have been comparatively less extensively investigated in this health area, however, in recent past there are some scattered reports of quinozolones as antibacterial agents.

These compounds were found not to exhibit any inhibitory effect as is evident from the antifungal activity data recorded in (Table 2). Five substituents were made in the quinozolone nucleus along with common substituent (C_6H_5 -) at position -2. As such these compounds do not warrant further investigation as far as antifungal activity is concerned. In order to explore the further possibility. Two attempts seem to be worthwhile. First, other pharmacophoric groups should be introduced at position 2 and 3 of the quinozolone nucleus and second, such compounds should be assayed against other fungal strains.

Conclusion

Amido/imidoalkyl groups have been introduced into the basic molecular architecture with the anticipation that the introduction of such groups might enhance the polarity character of new compounds thus facilitating their dissolution in water and other polar solvents making bioevaluation more feasible. Antibiotics belonging to the penicillin and cephalosporin classes are being uninterruptedly used in fighting the bacterial diseases since a longtime. However, emergence of resistance has become a problematic one and this emergence has increased worldwide during the past few years giving rise to an urgent need for new and more effective chemical candidate molecules. Since some of our compounds are showing promising antimicrobial activity selectivity and a majority of antibiotics and azoles group of antifungal in the arsenal of a physician are developing resistance, therefore in order to meet this challenge new agents are urgently required. These compounds need further investigations involving inclusion of more bacterial and fungal strains.

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Conflict of Interest

The author declares no potential conflicts of interest with respect to the research, authorship, and/or publication of this article. Informed consent was obtained for this publication.

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